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For acid solutions pOH is greater than 7, and for basic solutions pOH is less than 7. It is possible to have solutions in which pH or pOH are greater than 14 or less than 0 (i.e., negative). Note that the number of decimal places in a logarithm indicates the number of significant figures in the original number.

Chem 116 POGIL Worksheet - Week 9
Introduction to Acid ...

Model 3 — Conjugate Acid—Base Pairs
Base + 1-120 (I) HC031-(aq) 13. All acid—base reactions have two conjugate acid—base pairs. One conjugate acid—base

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pair in the reaction in Model 3 is H^+/OH^- . List the other acid—base pair in the reaction. 14. Why is FICO - considered the "acid" part of the pair in the reaction in Model 3? 15.

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