

Chapter 13 Gases An Introduction To Chemistry

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Chapter 13 – Gases 195 Exercise 13.3 – Equation Stoichiometry: Iron is combined with carbon in a series of reactions to form pig iron, which is about 4.3% carbon. $2C + O_2 \rightarrow 2CO$ $2CO + Fe_2O_3 \rightarrow 2Fe + 3CO_2$ $2CO + C \rightarrow 2C + CO_2$ Pig iron is easier to shape than pure iron, and the presence of carbon lowers its melting point

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Gas Model. • Gases are composed of tiny, widely-spaced particles. – For a typical gas, the average distance between particles is about ten times their diameter. Gas Model (cont.) • Because of the large distance between the particles, the volume occupied by the particles themselves is negligible (approximately zero).

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Chapter 13 (Gases) EXAMPLE USED *Temperature must be converted to kelvins. WHAT HAPPENS. VARIABLES INVOLVED* $P_1V_1/T_1 = P_2V_2/T_2$. Avogadro. n increases, V increases. None. V & n . Balloons of different size. Pile of books on a cylinder. Aerosol can in a woodstove. $PV = k'$ $V = k''T$. $P = k'''T$. $V = k''''n$. $P_1V_1 = P_2V_2$. $V_1/T_1 = V_2/T_2$. $P_1/T_1 = P_2/T_2$. $V_1/n_1 = V_2/n_2$. 2/1/2006. Volume (cubic inches)

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Section 13.6 Exercise 27.4 L of oxygen gas at 25.0 ° C and 1.30 atm, and 8.50 L of helium gas at 25.0 ° C and 2.00 atm were pumped into a tank with a volume of 5.81 L at 25 ° C. • Calculate the new partial pressure of oxygen. 6.13 atm • Calculate the new partial pressure of helium. 2.93 atm • Calculate the new total pressure of both gases. 9.06 atm

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Chapter 13 Highlights Liquids and gases are both fluids, since they flow, but the molecules of a liquid are farther apart than those of a gas and they have a set volume while gases do not. They both take the shape of their container. A common instrument used for measuring air pressure is a barometer.

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Chapter 13: Gas Mixtures - Gas Mixtures The discussions in this chapter are restricted to nonreactive ideal-gas mixtures. Those interested in real-gas mixtures are encouraged to study carefully ... | PowerPoint PPT presentation | free to view

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Introduction; 9.1 Gas Pressure; 9.2 Relating Pressure, Volume, Amount, and Temperature: The Ideal Gas Law; 9.3 Stoichiometry of Gaseous Substances, Mixtures, and Reactions; 9.4 Effusion and Diffusion of Gases; 9.5 The Kinetic-Molecular Theory; 9.6 Non-Ideal Gas Behavior; Key Terms; Key Equations; Summary; Exercises

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Figure 13.1 The black strip found on the back of credit cards and driver ' s licenses is a very thin layer of magnetic material with information stored on it. Reading and writing the information on the credit card is done with a swiping motion. The physical reason why this is necessary is called electromagnetic induction and is discussed in this chapter.

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Chapter 37 Solution-Gas-Drive Reservoirs Roger J. Steffensen, Amoco Production Co.* Introduction An oil reservoir is a solution-gas-drive reservoir if it un-dergoes primary depletion with the main reservoir energy supplied by the release of gas from the oil and the expan-sion of the in-place fluids as reservoir pressure drops.

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