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pressure is controlled through the placement of masses on the lid of the container, and temperature is controlled with an adjustable heat source.

Boyle's Law and Charles's



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and others in which the pressure remains fixed (Charles's Law). The pressure is controlled through the placement of masses on the lid of the container, and temperature is controlled with an

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adjustable heat source.

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Gizmo shows a container of  
gas. Inside, small purple

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spheres represent gas  
molecules. 1.

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Charles's Law Name 1. Get the Gizmo ready: Set the temperature ( $T$ ) to 300 K and the mass ( $m$ ) to 0 kg. A. With the temperature held constant at 300 K, use the Select mass slider to place

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Answer Key  
particles. Are they all moving at the same speed? No, the particles are moving at different speeds.

Student Exploration Gizmo-  
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...

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Name: Avery Miller Date: Jan  
16 2020 Chemistry B Unit 1  
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Exploration: Boyle's Law and  
Charles' Law Vocabulary:  
absolute zero, Boyle's law,  
Charles' law, Gay-Lussac's  
law, Kelvin scale, pressure



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Student Exploration: Boyle's  
Law and Charles's Law

Vocabulary: absolute zero,  
Boyle's law, Charles's law,  
Gay-Lussac's law, Kelvin

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Answer Key  
scale, pressure Prior  
Knowledge Question (Do this  
BEFORE using the Gizmo.) A  
small helium tank measures  
about two feet (60 cm) high.  
Yet it can fill over 50  
balloons! How can such a  
small tank contain enough

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Answer Key to fill so ...

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Gizmo Warm-up On the Boyle's Law and Charles' Law Gizmo™, check that the BOYLE'S LAW tab is selected. The Gizmo shows a container of gas; the little purple spheres represent molecules. 1.

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Gizmo : Lesson Info ...

Investigate the properties  
of an ideal gas by  
performing experiments in  
which the temperature is  
held constant (Boyle's Law),  
and others in which the

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pressure remains fixed  
(Charles' Law) .

## Boyle's Law And Charles Law Gizmo Answers Activity A

When the balloon is filled  
with helium, helium occupies  
much more space since there

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Answer Key  
is less pressure in the balloons than the tank. Gizmo Warm-up The Boyle's Law and Charles's Law Gizmo shows a container of gas. Inside, small purple spheres represent gas molecules. 1.

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S8 Charles and Boyles

Law.docx - Name Date Student

...

For the Boyle's Law and Charles' Law Gizmo, we merged the Boyle's Law and Charles' Law tabs into a single Simulation pane. This

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Answer Key  
was done because many users were confused by having to set the value of one variable before investigating the effect of the other. The temperature and mass sliders are now always active.

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Changes to Boyle's Law and  
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How Student Exploration:  
Boyle's Law and Charles' Law  
Sample answer: "Charles' law  
is a direct relationship: as  
temperature increases,

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Answer Key  
volume increases. Boyle's  
law is an inverse  
relationship: as pressure  
increases, volume decreases.  
Page 13/25. Get Free Charles  
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Key.



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## Charles And Boyles Law Gizmo Answer Key

Sample answer: "Charles' law is a direct relationship: as temperature increases, volume increases. Boyle's law is an inverse relationship: as pressure

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Answer Key  
increases, volume decreases.  
Activity C: Gay-Lussac's law.  
Get the Gizmo ready: On the  
SIMULATION pane, set  $T$  to 100  
K and  $m$  to 0 kg.

Activity B: Get the Gizmo  
ready: Charles' T m

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Charles' law and Boyle's law are two very important laws concerned with gases. These two laws can describe many properties of the ideal gases. These laws are widely used in fields such as chemistry, thermodynamics,

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Answer Key even military  
applications.

Difference Between Charles  
Law and Boyle law | Compare  
the ...

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**Vocabulary:** absolute zero,  
Boyle's law, Charles' law,  
Kelvin scale, pressure  
Prior Knowledge Question (Do this  
BEFORE using the Gizmo.) A  
small helium tank measures  
about two feet (60 cm) high.

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