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CHAPTER 9

SOLUTIONS MANUAL

35. Aqueous
solutions of
potassium iodide and
silver nitrate are
mixed, forming the
precipitate silver

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iodide. chemical

equation: $KI(aq)$

$AgNO_3(aq) + 0 KNO$

$3(aq) + AgI(s)$ complete

ionic equation: $K(aq) + I$

$(aq) + 3 Ag^+(aq) + NO_3^-$

$(aq) + 0 K^+(aq) + NO_3^-(aq)$

$AgI(s)$

Chemical

Reactions

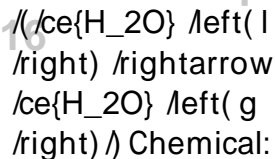
Chemical
Reactions

Solution. Physical:

boiling and melting

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are physical changes. When water boils, no bonds are broken or formed. The change could be written:



The dark grey nail changes color to form an orange flaky substance (the rust); this must be a

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30. Compare and
contrast sublimation
and evaporation. In

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both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

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