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Step 1: Note the ratio
of H^+ to OH^- is 1 to

1 Step 2: Substitute
values into the

equation $M_A \times V_A =$
 $M_B \times V_B$ $M_A(30. \text{ mL})$
 $= (1.2 \text{ M})(50. \text{ mL})$

Step 3: Solve for the
unknown M 12M

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