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~~Compounds | How~~

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relationship

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b. Inverse 3. One
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is to a. decrease
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increase volume

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b. Inverse 3. One
way to increase
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temperature b.
increase volume

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Gas Law Problems

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Steps to Solve any
Gas Law Problem:

- o Step 1: Write everything you are given in the problem.
- o Step 2: Which law do you want to use? (What remains constant?)
- o Step 3: Do your units match? If not, convert.
(Temperature must always be in

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Kelvin) o Step 4:
Plug in your values
and solve.

Proportional
Indirectly Directly
Directly

~~Gas Laws Notes
KEY 2015-16~~

Unit 2 Gas Laws
Test Review 1. e
tfml peatr user ci
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Answers and
volumes i 1. direct
2. inverse 2. If
pressure is
constant, the
relationship
between
temperature and
volume is a. direct
b. Inverse 3. One
way to increase
pressure on a gas
is to a. decrease
temperature b.

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REVIEW/PRACTICE
SHEET ANSWERS.

$$R = 8.31 \text{ (kPa)(L) / (mol)(K) = 62.36}$$

$$\text{(mmHg)(L) /}$$

$$\text{(mol)(K) = 0.082}$$

$$\text{(atm)(L) / (mol)(K)}$$

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Match each of the following statements/equations to the corresponding name:

Charles Law
 $P_1V_1 = \text{constant}$

Boyles Law
 $P_1V_1/T_1 = P_2V_2/T_2$

Combined gas equation
 $V_1/T_1 = \text{constant}$

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ANSWER SHEET~~

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The Results for Gas
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Key. Function
Worksheet. Gas
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Problems

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Key. Combined gas
law $(P_1 V_1)/T_1 =$
 $(P_2 V_2)/T_2$ (T
must be in Kelvin)

Ideal gas law: $PV =$
 nRT ($R = 0.0821$ L
atm/K.mol) The

Combined Gas Law

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~~Answer Key~~ Law -
dummies The
combined gas law
combines the three
gas laws: Boyle's
Law, Charles' Law,
and Gay-Lussac's
Law.

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Chemfiesta Answer
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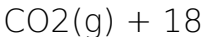
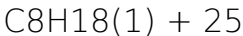
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Ideal Gas Law and

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Stoichiometry Use
the following
reaction to answer
the next few
questions: 2



$\text{H}_2\text{O}(\text{g})$ The above
reaction is the
reaction between
gasoline (octane)
and oxygen that
occurs inside

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Answer Key
automobile
engines.

~~Gas Laws STUDY
GUIDE Due:~~

~~February 12th~~

The three
fundamental gas
laws discover the
relationship of
pressure,
temperature,
volume and
amount of gas.

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Boyle's Law tells us that the volume of gas increases as the pressure decreases. Charles' Law tells us that the volume of gas increases as the temperature increases. And Avogadro's Law tell us that the volume of gas increases as the amount of gas

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Answer Key

increases. The ideal gas law is the combination of the three simple gas laws.

~~Gas Laws:~~

~~Overview~~

~~Chemistry~~

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5 Gas Law Answer

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~~Unit 5: Gas Laws~~
~~Mrs. Zuberbuehler~~
Balancing
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Worksheet.
Solubility Curve
Practice Problems

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Worksheet. Gas
Laws Worksheet
Answers. Function
Worksheet. Protons
Neutrons and
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Worksheet.

~~Electron
Configuration~~

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Answers ...~~

Practice Test: Gas
Laws. 11. Zinc
metal is added to
hydrochloric acid to
generate hydrogen
gas and is collected
over a liquid whose
vapor pressure is
the same as pure
water at 20.0°C (18
torr). The volume
of the mixture is

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1.7 L, and its total pressure is 0.810 atm.

~~Practice Test: Gas Laws~~

The formula of this law is as follows:
(8.4.6) $P V = n R T$.
In this equation, P is pressure, V is volume, n is amount of moles, and T is

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temperature. R is called the ideal gas law constant and is a proportionality constant that relates the values of pressure, volume, amount, and temperature of a gas sample.

~~8.4: Gas Laws~~
~~Chemistry~~
~~LibreTexts~~

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