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CHAPTER 15

REVIEW Acid-Base

Titration and pH

CHAPTER 15

REVIEW Acid-Base

Titration and pH

SHORT ANSWER

Answer the following
questions in the space

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provided 1 Calculate the following values without using a calculator 40 a The $[H^+]$ in a solution is $1 \times 10^{-4} M$ Calculate the pH
___; $1 \times 10^{-4} M$...

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continued 6. Three
students were asked to
determine the volume of
a liquid by a method of
their choosing. Each did

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three trials. The table below shows the results.

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Chemistry
Titration and pH

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SHORT ANSWER

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questions in the space

provided 1 Calculate the

following values without

using a calculator 40 a

The $[H^+]$ in a

solution is $1 \times 10^{-4} M$

Calculate the pH

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questions in the space
provided. 1. Calculate
the following values
without using a
calculator. 4.0 . a. The
[H. 3. 0+] in a solution
is $1 \cdot X \cdot 10^{-4} \cdot M$.

Calculate the pH. ____.;.

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1_X----'1:..:0'--_1_3-'-

M ~ ___ b. The pH of a solution is . 13.0.

Calculate the [H ...

Answers

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react with active metals

and release hydrogen

gas. • Acids react with

bases to produce $\text{SO}_2 +$

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H₂O H₂SO₃ SO₃ +

H₂O H₂SO₄ NO +

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Gases consist of large

numbers of tiny particles

that are far apart

relative to their size.

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for many people.

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